

# Chemistry If8766 Instructional Fair Inc Nuclear Decay Answers

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### Chemistry If8766 Instructional Fair Inc

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Chemistry IF8766 Olnstructlonal Fair, Inc STOICHIOMETRY: MOLE-MOLE PROBLEMS 1 N<sub>2</sub> +3H<sub>2</sub> Name How many moles of hydrogen are needed to completely react with two moles of nitrogen? 20 +302 Chemistry 'Instructional Fair, tnc

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**LE CHATELIER'S PRINCIPLE Name - Ms. Ose's Chemistry Website**

Chemistry IF8766 82 ©Instructional Fair, Inc Le Chatelier's Principle states that when a system at equilibrium is subjected to a stress, the system will shift its equilibrium point in order to relieve the stress Complete the following chart by writing left, right or none for equilibrium shift, and

**KMBT 654-20140224095350 - Berger's Chemistry Class**

©Instructional Fair, Inc 8 Decrease Temperature 9 Increase Pressure Decrease Pressure Chemistry IF8766 C LE CHATELIER'S PRINCIPLE CONTINUED 126 kcal + H<sub>2</sub>(g) Name 12 (g) 2Hl(g) Stress Add H<sub>2</sub> 2 Add 12 3 Add HI 4 Remove H 5 Remove I 6 Remove HI Increase Temperature 8

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Olnstructlonal Fair, Inc Chemistry IF8766 E CHATELIER'S PRINCIPLE Name Chatellor's Prlnclple states that when a system at equillbrlum Is subjecte a stress, the system will shift Its equillbrlum point In order to relieve the stress Complete tho following chart ...

**web1.tvusd.k12.ca.us**

©Instructional Fair, Inc 20 Chemistry IF8766 CHARLES' LAW Name charles' Law states that the volume of a gas varies directly with the Kelvin temperature, assuming that pressure is constant We use the following formulas:  $V \propto T$  or  $V_1/T_1 = V_2/T_2$

**www.manhassetsschools.org**

Instructional Fair, Inc 5 35 g of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  Determine the number of grams in each of the quantities below 1, 25 moles of NaCl 2 050 moles of  $\text{K}_2\text{SO}_4$  3, 170 moles of  $\text{KMnO}_4$  4, 025 moles of KCl e, ST 5 32 moles of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  Chemistry IF8766

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**newburyparkhighschool.net**

VAPOR PRESSURE AND BOILING 700 600 Name A liquid will boil when its vapor pressure equals atmospheric pressure Answer the questions following the graph

**Writing Formulas (KEY) (Criss-Cross Method)**

Writing Formulas (KEY) (Criss-Cross Method) Cl-CO 3-2 OH-SO 4-2 PO 4-3 NO 3-Na+ NaCl Na 2CO 3 NaOH Na 2SO 4 Na 3PO 4 NaNO 3 NH 4 + NH 4Cl (NH 4) 2CO 3 NH 4OH (NH 4) 2SO 4 (NH 4) 3PO 4 NH 4NO 3 K+ KCl K 2CO 3 KOH K 2SO 4 K 3PO 4 KNO 3 Ca+2 CaCl 2 CaCO 3 Ca(OH) 2 CaSO 4 Ca 3(PO 4) 2 Ca(NO 3) 2 Mg+2 MgCl 2 MgCO 3 Mg(OH) 2 MgSO 4 Mg 3(PO 4) 2 Mg(NO 3) 2 Zn+2 ...

**Balancing Chemical Equations Answer Key**

Balancing Chemical Equations -Answer Key Balance the equations below: 1)  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  2)  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$  3)  $2\text{NaCl} + 1\text{F}_2 \rightarrow 2\text{NaF} + 1\text{Cl}_2$  4)  $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$  5)  $1\text{Pb(OH)}_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{PbCl}_2$  6)  $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2(\text{SO}_4)_3$

**NAMING HYDROCARBONS Name**

NAMING HYDROCARBONS Name Name the compounds below according to the IUPAC naming system 1 H H H I I I H —C —C —C —H I I I H H H 5

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**SIGNIFICANT FIGURES Name**

Chemistry IF8766 9 Instructional Fair, Inc 59 CALCULATIONS USING SIGNIFICANT FIGURES Name ? When multiplying and dividing, limit and round to the least number of significant figures in any of the factors 11 Example 1 :  $230\text{ cm} \times 432\text{ cm} \times 19\text{ cm} = 188,784\text{ cm}^3$  11

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